## SEMESTER-TWO

### CHEMISTRY

# Class XII Sample Paper—2

Max. Marks: 50 Time Allowed: 90 minutes

#### General Instructions:

- (i) This question paper consists of 40 questions in 4 sections.
- (ii) Section A consists of 10 Objective type questions carrying 1 mark each.
- (iii) Section B consists of 10 Fill in the blanks type questions carrying 1 mark each.
- (iv) Section C consists of 10 True or False statement type questions carrying 1 mark each.
- (v) Section D consists of 10 Short answer type questions carrying 2 marks each.

#### Section A

Select and write one most appropriate option out of the four options given for each of the questions 1-10.

- 1. Which of the following metals can be extracted by smelting?
  - (a) Aluminium

(b) Magnesium

(c) Iron

- (d) None of these.
- **2.** In the froth-floatation process for benefication of the ores, the ore particles float because
  - (a) they are light
  - (b) their surface is not easily wetted by water
  - (c) they bear electrostatic charge
  - (d) they are insoluble.

- Which of the following benefication processes is used for the mineral bauxite?
  - (a) Froth floatation

(b) Leaching

(c) Liquation

(d) Magnetic separation.

Which type of the coal has the highest heating value? 4.

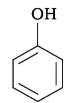
(a) Anthracite

(b) Bituminous

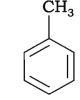
(c) Sub-bituminous

(d) Lignite

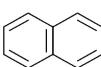
- Syngas is a mixture of 5.
  - (a) Hydrogen & Carbon monoxide
  - (b) Hydrogen & Carbon dioxide
  - (c) Helium & Carbon monoxide
  - (d) None of these
- **6.** Which of the following is true about ammonia?
  - (a) Ammonia is a colourless gas.
  - (b) Ammonia is lighter than air.
  - (c) Ammonia is insoluble in water.
  - (d) Ammonia liquefies on cooling under pressure.
- **7**. Which of the following is a non-benzoic compound?



(a) Phenol



(b) Toluene



Napthalene (c) (Fused rings)



(d) Tropolone

- Which of the following is an acidic oxide?
  - (a) CaO

(b) K<sub>2</sub>O

(c)  $CO_2$ 

(d) BaO

- ..... is something that occupies space and has mass.
  - (a) Matter

(b) Thought

(c) Love

(d) None of these

- **10.** How many base units are there?
  - (a) Three

(b) Five

(c) Seven

(d) None of these

#### Section B

Fill in the blanks with a suitable word for each of the questions 11–20.

- **11.** The process of removal of gaugue from ore is known as ......
- **12.** Aluminium is obtained from Al<sub>2</sub>O<sub>3</sub> by ..... reduction.
- **13.** In the metallurgical process for electrorefining of the metal, the anode is made of ..... metal.
- **14.** Slag is a product formed during smelting by combination of ..... and impurities.
- **15.** ..... is called oil of vitriol.
- **16.** ..... discovered chlorine.
- **17.** A reaction which cannot take place in the reverse direction is called the ...... reaction.
- **18.** ..... is the speed at which reactants are converted into products.
- 19. Liquid ammonia is used as a .............
- **20.** Hydrocarbons are made up of ...... and .............

#### **Section C**

State whether the following statements are true or false for each of the questions 21–30.

- 21. Subbituminous coal typically contains 35-45% carbon,
- **22.** The process of cooling coal in the absence of air is called the destructive distillation of coal.
- 23. Coal is primarily used as fuel to generate electric power.
- **24.** An unimportant use of syngas is as a **feedstock** for the production of an amazing number of chemicals.
- **25.** Synthesis gas (syngas in brief) is a gas mixture of predominantly CO and  $H_2O$ .
- **26.** In the **laboratory**, dinitrogen is prepared by heating aqueous solution of ammonium chloride with sodium nitrite.
- 27. Ammonia is hydride of nitrogen.
- **28.** Ammonia acts as a Lewis base due to the presence of a lone pair of electrons on nitrogen.

- **29.** The impure acid is generally yellow due to the presence of nitrogen dioxide as impurity.
- 30. Noble metals like gold and platinum react sharply with nitric acid.

#### Section D

Answer each of the questions 31-40.

- 31. What do you mean by alloy? Give some examples.
- **32.** What is cement. Write down its compostion.
- **33.** What is dry ice? List its use.
- **34.** Who discovered nucleus of an atom?
- **35.** List some uses of sulphur dioxide.
- **36.** What do you mean by sulphur?
- 37. What do you understand by halogens?
- **38.** What do you understand by electronegativity?
- **39.** Who discovered chlorine?
- **40.** Where does calcium sulphate occur?